

A Novel Undecametallic Iron(III) Cluster with an $S = {}^{11}l_2$ Spin Ground State

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The reaction of $[NEt_4]_2[Fe_2OCl_6]$ with sodium benzoate, 4,6dimethyl-2-hydroxypyrimidine (dmhp), and 1,1,1-tris(hydroxymethyl)ethane (H₃thme) gives the undecametallic compound $[NEt_4]$ - $[Fe_{11}O_4(O_2CPh)_{10}(thme)_4(dmhp)_2Cl_4]$. X-ray crystallography, EPR spectroscopy, bulk magnetic susceptibility studies, and lowtemperature single-crystal magnetic measurements were used to characterize the compound. Magnetic measurements indicate an $S = {}^{11}/_2$ ground state with the parameters g = 2.03 and D =-0.46 cm⁻¹. Single-crystal magnetic studies show hysteresis of molecular origin at T < 1.2 K with fast quantum mechanical tunneling at zero field.

Iron-oxo clusters have been investigated for many years for a variety of reasons. These range from biological models for the iron oxo core of the iron storage protein ferritin,¹ the study of biomineralization processes that form iron-oxo minerals,² and more recently for single-molecule magnetism (SMM) behavior. Although a variety of Mn-containing SMMs have been reported in recent years,³ fewer Fe SMMs are known.⁴⁻⁸ The requirement for SMM behavior is the

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combination of a large spin ground state with a negative *D* value, but there are relatively few Fe(III) clusters with large ground spin states: Fe4, S = 5;⁵ Fe4, S = 8;^{8,9} Fe8, S = 10;⁴ Fe10, S = 11;⁶ Fe7, $S = \frac{29}{2}$;¹⁰ and Fe19, $S = \frac{33}{2}$;¹¹ Following on from our recent success using the ligand 1,1,1-tris(hydroxymethyl)ethane (H₃thme) in Mn and Fe cluster chemistry,^{12,13} we herein report the synthesis, structure, and magnetic properties of a new undecametallic Fe cluster with an $S = \frac{11}{2}$ spin ground state.

Reaction of $[NEt_4]_2[Fe_2OCl_6]$ (1 equiv) with NaO₂CPh (2 equiv), dmhp (1 equiv), and H₃thme (1 equiv) in MeCN leads to the formation of $[NEt_4][Fe_{11}O_4(O_2CPh)_{10}(thme)_4(dmhp)_2Cl_4]$ 1¹⁴ in $\leq 20\%$ yield after 1 week. **1** (Figure 1) crystallizes in the monoclinic space group $P2_1/c$.

The core of **1** (Figure 2) consists of four fused butterfly $\{Fe_4O_2\}^{8+}$ motifs. The two butterflies in the center of the core (Fe1, Fe2, Fe3, Fe5, and symmetry equivalents) share a body Fe ion (Fe3) forming a planar central $\{Fe_7O_4\}^{13+}$ unit (Fe1–Fe3–Fe1A = 180.0°). The two peripheral but-

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- (14) Crystal data for 1: $C_{131}H_{130}Cl_4Fe_{11}N_{14}O_{38}$, M = 3264.64, monoclinic, space group P_{21}/c , a = 13.9889(9) Å, b = 21.5807(15) Å, c = 23.9145(16) Å, $\beta = 94.753(1)^\circ$, V = 7194.7(8) Å³, Z = 2, $D_c = 1.507$ g cm⁻³, $\mu = 1.229$ mm⁻¹ (Mo K α , $\lambda = 0.71073$ Å), T = 100(2) K, CCD diffractometer, 41276 measured, 14643 unique data ($R_{int} = 0.0408$), of which 11583 have $I > 2\sigma(I)$. R ($I > 2\sigma(I)$) = 0.0400, wR2 (on F^2 , all data) = 0.1105 with 922 refined parameters, goodnessof-fit = 1.028.

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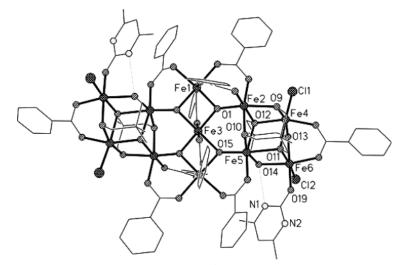


Figure 1. The structure of the complex in 1. Selected interatomic distances (Å): $Fe^{3+}-O$, 1.827(18)-2.182(18), $Fe^{3+}-Cl$, 2.266(8)-2.301(8).

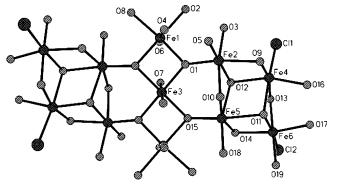
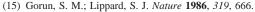


Figure 2. The core of the complex in 1.

terfly units (Fe2, Fe4, Fe5, Fe6, and symmetry equivalents) are nonplanar and are situated, respectively, above and below the plane of the central $\{Fe_7O_4\}^{13+}$ core. The wing tip Fe atoms (Fe2 and Fe6) lie respectively above and below the plane of the body Fe atoms Fe4 and Fe5. The four thme³⁻ ligands are found in the peripheral $\{Fe_4O_2\}$ units, with two oxygen atoms acting as μ_2 -bridges, linking the Fe atoms within the $\{Fe_4O_2\}$ unit as well as between the $\{Fe_4O_2\}$ and $\{Fe_7O_4\}^{13+}$ units. The third oxygen atom acts as a μ_3 -bridge within the $\{Fe_4O_2\}$ unit. The 10 PhCO₂⁻ ligands bridge in their familiar μ_2 -mode, while the Cl⁻ ions are all terminal. The neutral dmhp ligands are also terminally bonded, with the nitrogen atom protonated and hydrogen bonding to a μ_2 -oxygen of a thme³⁻ ligand (O14–N1, 2.796(8) Å).

{ M_4O_2 } butterfly units are common building blocks in the structures of Fe(III) and Mn(III) clusters. The { M_7O_4 }¹³⁺ subunit is also a common feature in larger clusters, including Fe17/19¹¹ and Fe11.¹⁵ In fact the Fe11 core is similar to a Mn11 cluster reported by Christou and co-workers, but in this case the peripheral units are [Mn4] cubes not butterflies.¹⁶ **1** is only the second undecametallic Fe complex reported;



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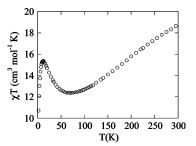


Figure 3. Thermal variation of $\chi_M T$ in the ranges 300–50 K at 0.2 T and 50–2 K at 0.05 T.

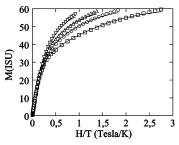


Figure 4. Magnetization vs H/T plot at $2(\Box)$, $3(\bigcirc)$, $4(\triangle)$, and $6(\bigtriangledown)$ K.

the original has no structural resemblance to **1**, being based on a pentacapped trigonal prism.

Solid state dc magnetization measurements were performed on 1 in the range 2-300 K in a field of 0.2 T between 300 and 50 K, and 0.05 T between 50 and 2K. (Figure 3). The room temperature $\chi_m T$ value of approximately 18.7 cm³ mol⁻¹ K decreases to a minimum of 12.4 cm³ mol⁻¹ K at 70 K, and then rises to a maximum of $15.4 \text{ cm}^3 \text{ mol}^{-1} \text{ K}$ at 12 K and then drops. This indicates that 1 has a nonzero spin ground state and that the ground spin state is higher than $\frac{9}{2}$. In order to obtain the ground state spin, magnetization data were collected in the ranges 2.0-6.0 K and 0.01-5.5 T (Figure 4). The fits were made simultaneously on the four different temperatures (2, 3, 4, and 6 K), assuming that only the ground state is populated, giving a best fit of S = $^{11}/_{2}$, g = 2.03, and D = -0.46 cm⁻¹. The maximum value of $\chi_{\rm M}T$ at low temperature (15.4 cm³ mol⁻¹ K) and the fact that the magnetization data at 2 and 6 K can be fit with one set of parameters strongly indicate that the ground state is

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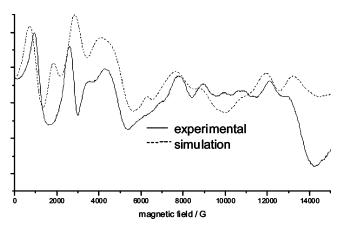


Figure 5. Q-band EPR spectrum of **1**. Experimental: v = 33.9250 GHz; T = 5 K, power = 1.9 mW; modulation amplitude = 10 G; Simulation: g = 2.03; D = -0.38 cm⁻¹; E = 0.055 cm⁻¹; $\Delta B = 1300$ G; Gaussian line shapes were used.

indeed $S = \frac{11}{2}$ and that the first excited state is rather high in energy. The previously reported Fe11 complex has an $S = \frac{1}{2}$ spin ground state.

The polycrystalline EPR spectrum of **1** at Q-band frequency and 5 K (Figure 5) confirmed both the spin state and magnitude and sign of the zero-field splitting deduced from the magnetization curves. Significantly, it was impossible to produce an adequate simulation of the spectrum without inclusion of rhombicity ($E \neq 0$). Given the nature of *S* and sign of *D*, measurements were taken to see if **1** acts as a SMM. Low-temperature (1.2–0.04 K) single-crystal magnetic measurements were performed on **1** using a micro-SQUID instrument equipped with three orthogonal fields, allowing the magnetic field to be scanned in all directions.

Below 1.2 K hysteresis loops are seen in magnetization vs field studies whose coercivities increase with decreasing temperature (Figure 6). A detailed study of the field sweep rate dependence of the hysteresis loops showed that the hysteresis at nonzero fields is due not to a phonon bottleneck but to slow relaxation because of the anisotropy barrier. Slow relaxation is seen at H = 0, but the presence of strong tunneling does not allow for a reliable Arrhenius plot. For all SMMs the barrier is reduced at H = 0 because of the presence of tunneling. Even half-integer spin systems, which in theory should not tunnel, do so because of coupling with the environment: dipolar coupling between molecules, hyperfine coupling, spin-spin cross relaxation, and other multibody quantum processes. For example for a system with $S = \frac{11}{2}$, D = -0.46 cm⁻¹, E = -0.055 cm⁻¹ and assuming an internal transverse field of approximately 10 mT, one can estimate a tunnel splitting of ca. 2.8×10^{-6} K and thus a tunnel probability of P = 0.68 for a sweep rate of 0.1 T s⁻¹, in good agreement with our measurements. This has also been observed in a [Mn4] cluster with $S = \frac{9}{2}$.¹⁷

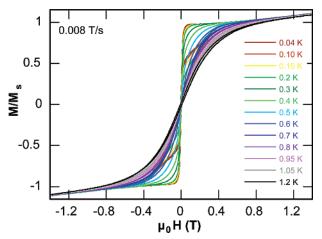


Figure 6. Magnetization of **1** (*M*) plotted as a fraction of maximum (M_s) vs applied magnetic field ($\mu_0 H$).

Magnetic studies of {Fe₄O₂}⁸⁺ butterflies indicate complexes with S = 0 spin ground states,¹⁸ in contrast to ${Mn_4O_2}^{8+}$ complexes which usually exhibit an S = 3 spin ground state.¹⁹ The difference has been explained by considering the different competing exchange interactions present: in $\{Mn_4O_2\}^{8+}$ complexes the body-body interaction is greater than the wing tip-body interaction, resulting in a "frustrated" spin alignment of the wing tip ions and an S =3 ground state. In ${Fe_4O_2}^{8+}$ complexes the body-wing tip exchange interaction dominates, resulting in an S = 0 spin ground state. 1 contains four such {Fe₄O₂} units fused together, creating a number of competing exchange interactions leading to the stabilization of an intermediate spin ground state of $S = \frac{11}{2}$. 1 is only the second Fe11 cluster reported (although various derivatives of the original Fe11 have been synthesized)²⁰ and is the first to exhibit an appreciable spin ground state. Combined with a negative Dvalue this leads to 1 being a new example of a SMM.

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Supporting Information Available: Crystallographic data in CIF format. This material is available free of charge via the Internet at http://pubs.acs.org.

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